

# FOUNDATIONS OF KINETIC THEORY

M. KAC  
CORNELL UNIVERSITY

## 1. Introduction

The basic equation of the kinetic theory of dilute monatomic gases is the famous nonlinear integro-differential equation of Boltzmann. In the simplest case when the molecules of the gas are hard spheres of diameter  $\delta$ , which are allowed to exchange energy only through elastic collisions, the Boltzmann equation assumes the form

$$(1.1) \quad \frac{\partial}{\partial t} f(\vec{r}, \vec{v}, t) + \vec{v} \cdot \nabla_{\vec{r}} f + \vec{X}(\vec{r}) \cdot \nabla_{\vec{v}} f = \frac{\delta^2}{2} \int d\vec{w} \int d\vec{l} \\ \cdot \{ f(\vec{r}, \vec{v} + (\vec{w} - \vec{v}) \cdot \vec{l}, t) f(\vec{r}, \vec{w} - (\vec{w} - \vec{v}) \cdot \vec{l}, t) - f(\vec{r}, \vec{v}, t) f(\vec{r}, \vec{w}, t) \} \\ \cdot |(\vec{w} - \vec{v}) \cdot \vec{l}|;$$

here  $f(\vec{r}, \vec{v}, t) d\vec{r} d\vec{v}$  is the average number of molecules in  $d\vec{r} d\vec{v}$  at  $\vec{r}, \vec{v}$ ,  $\nabla_{\vec{r}} f$  the gradient of  $f$  with respect to  $\vec{r}$ ,  $\nabla_{\vec{v}} f$  the gradient of  $f$  with respect to  $\vec{v}$ ,  $\vec{l}$  a unit vector and  $d\vec{l}$  the surface element of the unit sphere.  $\vec{X}(\vec{r})$  is an outside force (for example, gravity) acting on a particle at  $\vec{r}$ . If the gas is enclosed in a container of volume  $V$  and if there are no exterior forces ( $\vec{X}(\vec{r}) \equiv 0$ ) we can set

$$(1.2) \quad f(\vec{r}, \vec{v}, t) = \frac{n}{V} f(\vec{v}, t),$$

where  $n$  is the total number of molecules, and note that it will be a solution of (1.1) if  $f(\vec{v}, t)$  is a solution of the reduced Boltzmann equation

$$(1.3) \quad \frac{\partial}{\partial t} f(\vec{v}, t) = \frac{n\delta^2}{2V} \int d\vec{w} \int d\vec{l} \{ f(\vec{v} + (\vec{w} - \vec{v}) \cdot \vec{l}, t) f(\vec{w} - (\vec{w} - \vec{v}) \cdot \vec{l}, t) \\ - f(\vec{v}, t) f(\vec{w}, t) \} |(\vec{w} - \vec{v}) \cdot \vec{l}|.$$

Equation (1.3) governs the temporal evolution of the velocity distribution while the spatial distribution remains uniform.

If the molecules are not hard spheres but are considered as centers of force,

$$(1.4) \quad \frac{\delta^2}{2} |(\vec{w} - \vec{v}) \cdot \vec{l}|$$

has to be replaced by an expression depending on the nature of the force.

The most famous example is that of a Maxwell gas in which the molecules are

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